I-63

REACTIONS OF PENTAFLUOROSULFANYL ISOCYANATE WITH NR₃ AND PR₃ COMPOUNDS; PREPARATION AND CHARACTERIZATION OF NOVEL COMPOUNDS SF₂NC(O)NHC(O)OCH₃ AND 1,4-SF₅NHC(O)OC₆H₄OH

Alan F. Clifford*, Joseph S. Thrasher, Narayan S. Hosmane and Matthew Sullivan

Chemistry Dept., Virginia Polytechnic Institute & State Univ., Blacksburg, VA 24061 (U.S.A.)

The reaction of Pentafluorosulfanyl Isocyanate, SF_5NCO , with tertiary amines has led to 1:1 adducts which have been shown to be zwitterionic in nature by infrared and NMR spectroscopy.

These adducts are far less thermally stable than the analogous fluorosulfonyl derivatives recently reported by Appel and Montenarch 1 . The reaction of SF5NCO with triphenyl-phosphine gave some evidence for the zwitterionic derivative, but the product could not be isolated even when SF5NCO was used in excess. Recently we have also isolated

|| || F₂S=N-C-NH-C-O-CH₃ from the reaction of SF₂NC(0)NCO with methanol, and F₅S-NH-C-O-OH from the reaction of SF₅NCO with hydroquinone. Both of these novel compounds were unexpected. \bigcirc \bigcirc \bigcirc \bigcirc \bigcirc \bigcirc The physical properties of SF₅NC(0)NC₅H₅,SF₅NC(0)N(C₂H₅)₃, SF₂NC(0)NHC(0)OCH₃, and 1,4-SF₅NHC(0)OC₆H₄OH, prepared during this study, will be discussed in detail.

1 R. Appel and M. Montenarch, Chem. Ber., <u>110</u>, 2368(1977).

I-64

SOME REACTIONS OF FLUOROIMIDOTETRAFLUOROSULFUR

Brian A. O'Brien and Darryl D. DesMarteau

Department of Chemistry, Kansas State University, Manhattan, KS 66506 (U.S.A.)

Fluoroimidotetrafluorosulfur undergoes a number of reactions with strong electrophiles, involving addition to the sulfur-nitrogen double bond. It also reacts readily with fluoride ion, forming the reactive nucleophile SF_5NF .

Chlorine(I) fluorosulfate, bromine(I) fluorosulfate, and peroxydisulfuryl fluoride add to F4S=NF, forming the respective <u>cis</u> adducts. In the presence of KF, F4S=NF reacts readily with Br2, forming SF5NBrF, and with acyl fluorides, forming RCONFSF5. The SF5NFT ion also reacts with F5C=NF to form (SF5NF)FC=NF, which is isomerized to the unusual azo compound F5S-N=N-CF3 in the presence of CsF.

Self-reaction of F4S=NF in the presence of KF does not form the expected dimer $(SF_5NF)F_3S=NF$. Instead, extensive decomposition is observed, along with a low yield of the unusual amine $(SF_5)_2NF$.

The structures and possible mechanisms for formation of the various products will be discussed.